Questions from the "PAL Prüfungsbuch" Application of Chromatographic Methods All tasks are to be scored with 10 to 0 points

The following information correspond to the IUPAC recommendations "Nomenclature for Chromatography" and serve as a formulary for solution of the subsequent tasks.

$$\bar{\mathbf{u}} = \frac{L}{t_M} \qquad \qquad k' = \frac{t_R - t_M}{t_M} = \frac{t'_R}{t_M} \qquad \qquad \alpha = \frac{t'_R (Peak b)}{t'_R (Peak a)}$$

$$R_s = \frac{1.177 \times (t_{R2} - t_{R1})}{w_{h1} + w_{h2}} \quad \text{or} \quad R_s = \frac{2 \times (t_{R2} - t_{R1})}{w_{b1} + w_{b2}}$$

$$N = 5.545 \times ({t_R/W_h})^2$$
 or $N = 16 \times ({t_R/W_b})^2$

$$H = \frac{L}{N} \qquad MF = \frac{m_{Kal-STD} \times A_{ISTD}}{m_{ISTD} \times A_{Kal-STD}} \qquad w_{PR} = \frac{MF \times m_{ISTD} \times A_{Pr}}{A_{ISTD} \times m_{Pr}}$$

$$\beta_{Pr} = \frac{\beta_{STD} \times A_{Pr}}{A_{AUFG} - A_{Pr}}$$

t _M	Hold-up time (column dead time)	t _R	Retention time
ť _R	Reduced retention time	\mathbf{W}_{b}	Peak width at base line
\mathbf{W}_{h}	Peak width at half height	$\mathbf{m}_{\text{KAL-STD}}$	Mass of calibration standard
m _{ISTD}	Mass of internal standard	m _{Pr}	Mass os sample
L	Length of the chromatographic system	A _{Pr}	Peak area of the sample peak
b _{Pr}	Mass concentration in sample	A_{AUFG}	Peak area of spiked sample peak
b _{std}	Mass concentration of standard		



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1:													
A sample consists of													
Water													
 Ethen-1,2-diol (glycol) 													
 Propane-1,2,3-triol (glycerin) 													
in approx. equal parts)													
1. Suggest a chromatographic method to separate the mixture													
Explain your selection													
3. Choose a detector which is suitable for the quantitaive determination of all three analytes													

2	:

In a preliminary HPLC investigation it was found that parts of the sample matrix disrupt the chromatographic separation of the analytes by superposition of peaks. Name two measures that can be taken so that the analytes nevertheless can be faultlessly quantified. Lifelong Chemlab ¥*

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3:

The pictured chromatogram was taken with a 12.5 cm long HPLC-column. The inlet from the injector to the column is 2.5 cm, the outlet from the column to the detector is also 2.5 cm. A hold-up time marker was added to the two component sample. Calculate the following parameters:

- 1) Reduced retention time of peak number 2
- 2) Average linear flow velocity
- 3) Resolution between peak 2 and 3
- 4) Theoretical plate number regarding peak 3







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5:

- 1. Give a short description of the principle of a chromatographic analysis with an internal standard
- 2. In which case the method of the internal standard must be used? Name **two** major reasons.





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7: In a

- Gas chromatography (with thermal conductivity detector) or
- HPLC (with UV-detector)
- the column length is increased.

Explain how this affects the parameters that are specified in the table

Note:

Fill in the predefined answers given below **either** for gas chromatography **or** HPLC:

increases

decreases

remains constant

Hold-up time <i>t</i> _M	Retention time $t_{\rm R}$	Theoretical plate number N	Plate height H	Peak area A

8:

Explain the terms given in the table below.

Term	Explanation
Selectivity coefficient α = 1	
Resolution $R_{\rm s}$ = 1.5	
RP 18	
Headspace	
Permeability	
	L



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10: In a HPLC-two-point calibration the following data were determined: Weighed portion of analyte Peak area 2.00 mg 23400 counts 4.00 mg 44510 counts 2.50 mg pure analyte was added to the pure sample matrix without analyte. The mixture was then analyzed. A peak area of 28400 counts was obtained. Calculate the recovery rate RR (in %).



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