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## Quantification of salicylic acid

## Chemicals

- Salicylic acid
- Sodium hydroxide
- Sodium chloride
- Water, dest.


## Materials

- Measuring flask
- Volumetric pipette
- Measuring pipette
- Burettes
- UV/VIS-Photometer
- Quartz cuvettes
- Pipetting aid
- Fine jet washing bottle
- Graph paper or PC (spreadsheet programm)
- Precision balance


## Safety instructions



- wear safety goggles

- wear adequate safety gloves

| Salicylic acid | Sodium hydroxide |
| :---: | :---: |
| - H302, H318 <br> - P280, P305+P351+P338, P309+P310 <br> - HAZARD!! | - H314 <br> - P280 P301+P330+P331 P305+P351+P338 P309+P310 |
| $\stackrel{\bullet}{ }$ | - HAZARD!! |
|  | V |

## Experimental procedure

- Measuring wave length
- Molar extinction coefficient
- Linearity of method
- Cuvette
- Blank solution
- Molar mass salicylic acid
- Stock solution

298 nm
$\varepsilon=3500 \mathrm{~L} /(\mathrm{mol} \cdot \mathrm{cm})$
$\mathrm{E}=0.1$ bis 0.9
1 cm
deionized water
$\mathrm{M}=138.12 \mathrm{~g} / \mathrm{mol}$
solve 250 mg salicylic acid in 30 mL sodium hydroxid
in 1000 mL -measuring flask
4 hours
minimum 5

- Stability of solutions
- Quantity of calibration levels www.eu-chemlab.eu

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- Design of a dilution strategy
- Preparation of the calibration solutions as well as their photometric measurement
- Dilution of the sample and measurement
- Preparation of a calibration curve and determination of the sample concentration


## Waste disposal:

- Dispose of all wastes in the container for basic solutions


## Analysis:

- Calculation of the mass of salicylic acid in mg of the given sample

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## Preparation list

## Chemicals:

- Salicylic acid
approx. 0.5 g
- Sodium hydroxide
approx. 10 g
- Sodium chloride
approx. 100 mg


## Materials:

- Measuring flask
- Weighing scoop
- UV/VIS-Photometer
- Quartz cuvettes
- Volumetric pipettes
- Beaker glasses
- Measuring pipettes
- Spatula
- Measuring cylinder
- Burettes
- Pipetting aid
- Fine jet washing bottle
- Drying oven
- Precision balance
- Graph paper or PC (spreadsheet programm)

Preparation of the sample solution:

- Weigh $100-150 \mathrm{mg}$ salicylic acid in 100 mL -beaker glass and solve with 20 mL sodium hydroxid solution $\mathrm{w}=10 \%$. Transfer with deionized water in 100 mL -measuring flask and fill up

